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(54) **2-METHYLENE-19-NOR-20(S)-1 α -HYDROXY-BIS-HOMO-PREGNACALCIFEROL IN CRYSTALLINE FORM**

(52) **U.S. Cl.** **514/167; 552/653**
(58) **Field of Search** **514/167; 552/653**

(75) **Inventors:** **Hector F. DeLuca**, Deerfield, WI (US);
James B. Thoden, Madison, WI (US);
Hazel M. Holden, Fitchburg, WI (US);
Margaret Clagett-Dame, Deerfield, WI (US);
Sumithra Gowlugari, Madison, WI (US);
Pawel Grzywacz, Madison, WI (US)

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Primary Examiner—Alton N. Pryor
(74) *Attorney, Agent, or Firm*—Andrus, Scales, Starke & Sawall, LLP

(73) **Assignee:** **Wisconsin Alumni Research Foundation**, Madison, WI (US)

(57) **ABSTRACT**

(*) **Notice:** Subject to any disclaimer, the term of this patent is extended or adjusted under 35 U.S.C. 154(b) by 131 days.

A method of purifying 2-methylene-19-nor-20(S)-1 α -hydroxy-bis-homo-pregnacaciferol to obtain 2-methylene-19-nor-20(S)-1 α -hydroxy-bis-homo-pregnacalCIFEROL in crystalline form. The method includes the steps of boiling a solvent such as acetone under inert atmosphere, dissolving a product containing 2-methylene-19-nor-20(S)-1 α -hydroxy-bis-homo-pregnacalCIFEROL to be purified in the solvent, cooling the solvent and dissolved product below ambient temperature for a sufficient amount of time to form a precipitate of 2-methylene-19-nor-20(S)-1 α -hydroxy-bis-homo-pregnacalCIFEROL crystals, and recovering the 2-methylene-19-nor-20(S)-1 α -hydroxy-bis-homo-pregnacalCIFEROL crystals.

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Related U.S. Application Data

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(51) **Int. Cl.**⁷ **A61K 31/59; C07C 401/00**

9 Claims, 1 Drawing Sheet

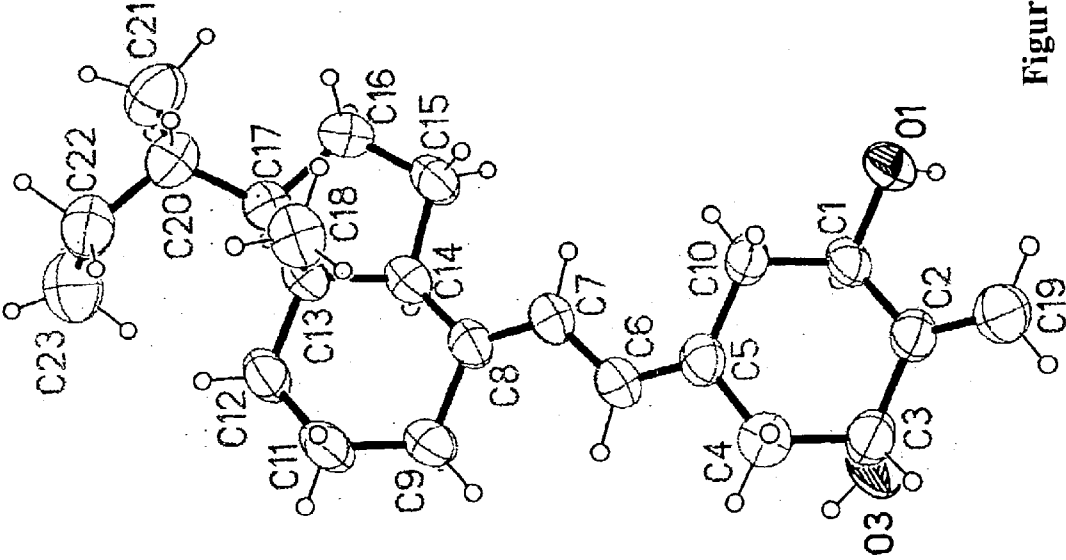


Figure 1

**2-METHYLENE-19-NOR-20(S)-1 α -HYDROXY-
BIS-HOMO-PREGNACALCIFEROL IN
CRYSTALLINE FORM**

**CROSS REFERENCE TO RELATED
APPLICATIONS**

This application is based on and claims priority from U.S. Provisional Patent Application No. 60/341,138 filed Dec. 13, 2001.

BACKGROUND OF THE INVENTION

The present invention relates to purification of organic compounds, and more particularly to the purification of 2-methylene-19-nor-20(S)-1 α -hydroxy-bis-homo-pregnacalCIFerol (referred to herein as 2MBP and/or 2-MbisP) by preparing it in crystalline form.

Purification of organic compounds, especially those designated for pharmaceutical use, is of considerable importance for chemists synthesizing such compounds. Preparation of the compound usually requires many synthetic steps and, therefore, the final product can be contaminated not only with side-products derived from the last synthetic step of the procedure but also with compounds that were formed in previous steps. Even chromatographic purification, which is a very efficient but relatively time-consuming process, does not usually provide compounds which are sufficiently pure to be used as drugs.

Depending on the method used to synthesize 1 α -hydroxyvitamin D compounds, different minor undesirable compounds can accompany the final product. Thus, for example, if direct C-1 hydroxylation of 5,6-trans geometric isomer of vitamin D is performed, followed by SeO₂/NMO oxidation and photochemical irradiation [see Andrews et al., *J. Org. Chem.* 51, 1635 (1986); Calverley et al., *Tetrahedron* 43, 4609 (1987); Choudry et al., *J. Org. Chem.* 58, 1496 (1993)], the final 1 α -hydroxyvitamin D product can be contaminated with 1 β -hydroxy- as well as 5,6-trans isomers. If the method consists of C-1 allylic oxidation of the 4-phenyl-1,2,4-triazoline-3,5-dione adduct of the previtamin D compound, followed by cycloreversion of the modified adduct under basic conditions [Nevinckx et al., *Tetrahedron* 47, 9419 (1991); Vanmaele et al., *Tetrahedron* 41, 141 (1985) and 40, 1179 (1991); Vanmaele et al., *Tetrahedron Lett.* 23, 995 (1982)], one can expect that the desired 1 α -hydroxyvitamin can be contaminated with the previtamin 5(10),6,8-triene and 1 β -hydroxy isomer. One of the most useful C-1 hydroxylation methods, of very broad scope and numerous applications, is the experimentally simple procedure elaborated by Paaren et al. [see *J. Org. Chem.* 45, 3253 (1980) and *Proc. Natl. Acad. Sci. U.S.A.* 75, 2080 (1978)]. This method consists of allylic oxidation of 3,5-cyclovitamin D derivatives, readily obtained from the buffered solvolysis of vitamin D tosylates, with SeO₂/t-BuOOH and subsequent acid-catalyzed cycloreversion to the desired 1 α -hydroxy compounds. Taking into account this synthetic path it is reasonable to assume that the final product can be contaminated with 1 α -hydroxy epimer, 5,6-trans isomer and the previtamin D form. 1 α -hydroxyvitamin D₄ is another undesirable contaminant found in 1 α -hydroxyvitamin D compounds synthesized from vitamin D₂ or from ergosterol. 1 α -hydroxyvitamin D₄ results from C-1 oxidation of vitamin D₄, which in turn is derived from contamination of the commercial ergosterol material. Typically, the final product may contain up to about 1.5% by weight 1 α -hydroxyvitamin

D₄. Thus, a purification technique that would eliminate or substantially reduce the amount of 1 α -hydroxyvitamin D₄ in the final product to less than about 0.1–0.2% would be highly desirable.

The vitamin D conjugated triene system is not only heat- and light-sensitive but it is also prone to oxidation, leading to the complex mixture of very polar compounds. Oxidation usually happens when a vitamin D compound has been stored for a prolonged time. Other types of processes that can lead to a partial decomposition of vitamin D compounds consist of the some water-elimination reactions; their driving force is allylic (1 α -) and homoallylic (3 β -) position of the hydroxy groups. The presence of such above-mentioned oxidation and elimination products can be easily detected by thin-layer chromatography. Thus, for example, using pre-coated aluminum silica sheets [with UV indicator; from EM Science (Cherry Hill, N.J.)] and solvent system hexane-ethyl acetate (4:6), the spot of 1 α -OH-D₂ (R_f 0.27) and its elimination products (R_f's ca. 0.7–0.9) are visible in ultraviolet light. Also, after spraying with sulfuric acid and heating, an additional spot can be visualized (R_f 0), derived from oxidation products.

Usually, all 1 α -hydroxylation procedures require at least one chromatographic purification. However, even chromatographically purified 1 α -hydroxyvitamin D compounds, although showing consistent spectroscopic data, suggesting homogeneity, does not meet the purity criteria required for therapeutic agents that can be orally, parenterally or transdermally administered. Therefore, it was evident that a suitable method of purification of 2MBP is required.

SUMMARY OF THE INVENTION

The present invention relates to a method of purifying 2MBP by means of crystallization to obtain 2MBP in crystalline form. The solvent plays a crucial role in the crystallization process, and is typically an individual liquid substance or a suitable mixture of different liquids. For crystallizing 2MBP, the most appropriate solvent and/or solvent system is characterized by the following factors:

- (1) low toxicity;
- (2) low boiling point;
- (3) significant dependence of solubility properties with regard to temperature (condition necessary for providing satisfactory crystallization yield); and
- (4) relatively low cost.

It was found that highly apolar solvents (e.g. hydrocarbons) were not suitable due to the low solubility of 2MBP in them. Quite the reverse situation occurred in highly polar solvent media (e.g. alcohols), in which 2MBP showed too high solubility. Therefore, it was concluded that for the successful crystallization of 2MBP, a solvent of medium polarity is required. Interestingly, hexane, so frequently used for crystallization purposes, was found less suitable for crystallization of 2MBP. However, it was found that an individual solvent, namely acetone, was most useful for the crystallization of 2MBP. In addition, it is believed that solvent systems utilizing ethyl acetate, or isopropanol, or chloroform, or dichloromethane, or diethyl ether would also perform well. These solvents are all very easy to remove by evaporation or other well known methods. In all cases the crystallization process occurred easily and efficiently; and the precipitated crystals were sufficiently large to assure their recovery by filtration.

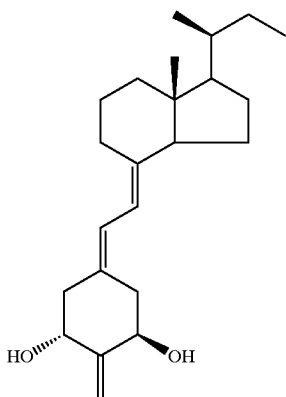
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BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 is an illustration of the three dimensional structure of 2MBD as defined by the atomic positional parameters discovered and set forth herein.

DETAILED DESCRIPTION OF THE INVENTION

The present invention provides 2-methylene-19-nor-20 (S)-1 α -hydroxy-bis-homo-pregnacalciferol (2MBP) in crystalline form, a pharmacologically important compound, characterized by the formula shown below:



The present invention also provides a valuable method of purification of 2MBP. The purification technique involves obtaining the 2MBP product in crystalline form by utilizing a crystallization procedure wherein the 2MBP material to be purified is dissolved using as the solvent a single solvent system comprised of acetone, ethyl acetate, isopropanol, chloroform, dichloromethane, or diethyl ether. The preferred solvent is acetone. Thereafter, the solvent can be removed by evaporation, with or without vacuum, or other means as is well known. The technique can be used to purify a wide range of final products containing 2MBP obtained from any known synthesis thereof, and in varying concentrations, i.e. from microgram amounts to kilogram amounts. As is well known to those skilled in this art, the amount of solvent utilized should be minimized and/or adjusted according to the amount of 2MBP to be purified.

The usefulness and advantages of the present crystallization procedures is shown in the following specific Examples 1 and 2. After crystallization, the precipitated material was observed under a microscope to confirm its crystalline form. Additionally, X-ray diffraction analysis was performed. Yields of crystallizations were high and the obtained crystals showed a relatively sharp melting point of 162–164° C.

The described crystallization process of the synthetic 2MBP product represents a valuable purification method, which can remove not only some side products derived from the synthetic path, but, moreover, concomitant 1 α -hydroxyvitamin D₄. Such impurity is the result of the contamination of natural ergosterol with its 22,23-dihydro analog.

Crystallization of 2MBP

EXAMPLE 1

Crystallization from Acetone

(a) 2MBP product to be purified may be dissolved in boiling acetone (1.2 mL, Aldrich) under argon atmosphere, left at room temperature (68° F.) for a few hours (1–3 hrs)

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and then in a refrigerator (35–45° F.) overnight (8–12 hrs). The precipitated crystals should be filtered off, washed with a small volume of a cold (0° C.) acetone and dried.

(b) These crystals of 2MBP may then be recrystallized with acetone (0.5 mL) as described in Example 1(a). The precipitated crystals have a relatively sharp melting point of 162–164° C., and were observed under a microscope to confirm their crystalline form.

EXAMPLE 2

Experimental

A colorless plate-shaped crystal of dimensions 0.40×0.25×0.03 mm was utilized for the structural analysis. Intensity data were collected with a Siemens HISTAR area detector mounted on a platform goniometer using Cu K α radiation ($\lambda=1.54178$ Å) focused with Goebel optics. The crystal was mounted on a thin nylon fiber with vacuum grease and data were collected at ambient temperature, 293° K. The intensity data were measured as a series of ω oscillation frames each of 0.25°; these were comprised of 5 scans (500 images each, 40 sec./image) at detector 2 θ of –85° and 2 scans (500 images each, 20 sec./image) at detector 2 θ of –40°. The detector was operated in 512×512 mode and was positioned 6.0 cm from the sample. Coverage of unique data was 95.4% complete to 58.88 degrees in θ . The first 50 frames were repeated at the end of data collection and showed essentially no decay of the crystal diffraction during the data collection. A total of 6917 data was measured in the range 4.38< θ <58.88°. The data were merged and scaled to form a set of 2697 independent data with R(int)=0.0562 using SAINT (Bruker-AXS, Inc.).

The orthorhombic space group P2(1)2(1)2(1) was determined by systematic absences and statistical tests and verified by subsequent refinement. The structure was solved by direct methods and refined by full-matrix least-squares methods on F² using SHELXL-97 (G. M. Sheldrick, 1994; SHELXTL Version 5 Reference Manual, Bruker-AXS, Inc.). Hydrogen atom positions were initially located in Fourier difference maps, but refined by a riding model with idealized geometries. Non-hydrogen atoms were refined with anisotropic displacement parameters. A total of 229 parameters were refined against 0 restraints and 2697 data to give $wR(F^2)=0.1369$ and $S=1.088$ for weights of $w=1/[s^2(F^2)+(0.0824 P)^2]$, where $P=[F_o^2+2F_c^2]/3$. The final R(F) was 0.0544 for the 2697 observed, $[F>2s(F)]$, data. The final difference map had maxima and minima of 0.126 and –0.153 e/Å³, respectively.

The three dimensional structure of 2MBP as defined by the following physical data and atomic positional parameters described and calculated herein is illustrated in FIG. 1.

TABLE I

Crystal data and structure refinement for 2-MbisP.	
Empirical formula	C23 H36 O2
Formula weight	344.52
Temperature	293(2) K
Wavelength	1.54178 Å
Crystal system, space group	Orthorhombic, P2(1)2(1)2(1)
Unit cell dimensions	a = 6.6320(13) Å $\alpha = 90^\circ$ b = 15.514(3) Å $\beta = 90^\circ$ c = 20.194(4) Å $\gamma = 90^\circ$
Volume	2077.7(7) Å ³
Z, Calculated density	4, 1.101 Mg/m ³
Absorption coefficient	0.520 mm ⁻¹
F(000)	760
Crystal size	0.40 × 0.25 × 0.03 mm
Theta range for data collection	4.38 to 58.88°

TABLE I-continued

Crystal data and structure refinement for 2-MbisP.	
Limiting indices	-7 ≤ h ≤ 5, -12 ≤ k ≤ 17, -22 ≤ l ≤ 21
Reflections collected/unique	6917/2697 [R _{int} = 0.0562]
Completeness to theta = 25.00	95.4%
Refinement method	Full-matrix least-squares on F ²
Data/restraints/parameters	2697/0/229
Goodness-of-fit on F ²	1.088
Final R indices [I > 2 sigma(I)]	R1 = 0.0544, wR2 = 0.1369
R indices (all data)	R1 = 0.0622, wR2 = 0.1453
Largest diff. peak and hole	0.126 and -0.153 e/Å ⁻³

TABLE 2

Atomic coordinates (×10⁴) and equivalent isotropic displacement parameters (Å² × 10³) for 2-MbisP. U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	U(eq)
O(1)	1842(3)	3525(1)	5011(1)	62(1)
O(3)	7954(4)	3113(1)	4476(2)	78(1)
C(1)	3883(5)	3729(2)	4852(2)	50(1)
C(2)	4569(5)	3359(2)	4201(2)	58(1)
C(3)	6698(5)	3569(2)	4030(2)	61(1)
C(4)	7074(6)	4537(2)	4057(2)	59(1)
C(5)	6301(5)	4937(2)	4686(2)	48(1)
C(6)	7497(5)	5433(2)	5059(2)	52(1)
C(7)	7012(5)	5872(2)	5673(2)	53(1)
C(8)	8248(5)	6371(2)	6020(2)	50(1)
C(9)	10390(5)	6589(2)	5828(2)	65(1)
C(10)	4161(5)	4704(2)	4851(2)	51(1)
C(11)	10739(6)	7562(2)	5809(2)	70(1)
C(12)	10108(6)	8006(2)	6452(2)	61(1)
C(13)	7933(5)	7806(2)	6640(1)	47(1)
C(14)	7696(5)	6814(2)	6661(2)	50(1)
C(15)	5663(6)	6668(2)	6982(2)	62(1)
C(16)	5453(5)	7416(2)	7474(2)	59(1)
C(17)	7251(5)	8038(2)	7353(2)	51(1)
C(18)	6466(6)	8205(2)	6135(2)	64(1)
C(19)	3405(7)	2898(3)	3811(2)	97(2)
C(20)	6703(6)	8967(2)	7542(2)	65(1)
C(21)	6012(8)	9017(3)	8269(2)	92(2)
C(22)	8368(7)	9636(2)	7418(2)	89(1)
C(23)	10284(9)	9512(3)	7809(3)	116(2)

TABLE 3

Bond lengths [Å] and angles [°] for 2-MbisP.

O(1)—C(1)	1.426(4)	O(3)—C(3)	1.416(4)
C(1)—C(2)	1.505(5)	C(1)—C(10)	1.524(4)
C(2)—C(19)	1.314(5)	C(2)—C(3)	1.490(5)
C(3)—C(4)	1.522(4)	C(4)—C(5)	1.504(5)
C(5)—C(6)	1.335(4)	C(5)—C(10)	1.502(5)
C(6)—C(7)	1.452(4)	C(7)—C(8)	1.327(4)
C(8)—C(9)	1.511(5)	C(8)—C(14)	1.511(4)
C(9)—C(11)	1.527(5)	C(11)—C(12)	1.530(5)
C(12)—C(13)	1.523(5)	C(13)—C(18)	1.539(5)
C(13)—C(14)	1.548(4)	C(13)—C(17)	1.553(4)
C(14)—C(15)	1.513(5)	C(15)—C(16)	1.534(5)
C(16)—C(17)	1.553(5)	C(17)—C(20)	1.535(5)
C(20)—C(22)	1.536(6)	C(20)—C(21)	1.540(5)
C(22)—C(23)	1.508(7)		
O(1)—C(1)—C(2)	113.5(2)	O(1)—C(1)—C(10)	109.6(2)
C(2)—C(1)—C(10)	109.9(3)	C(19)—C(2)—C(3)	122.5(3)
C(19)—C(2)—C(1)	123.6(3)	C(3)—C(2)—C(1)	113.9(3)
O(3)—C(3)—C(2)	107.5(3)	O(3)—C(3)—C(4)	111.9(3)
C(2)—C(3)—C(4)	111.3(3)	C(5)—C(4)—C(3)	112.5(3)

TABLE 3-continued

Bond lengths [Å] and angles [°] for 2-MbisP.				
5	C(6)—C(5)—C(10)	125.1(3)	C(6)—C(5)—C(4)	120.7(3)
	C(10)—C(5)—C(4)	114.2(3)	C(5)—C(6)—C(7)	128.3(3)
	C(8)—C(7)—C(6)	126.0(3)	C(7)—C(8)—C(9)	125.1(3)
	C(7)—C(8)—C(14)	124.6(3)	C(9)—C(8)—C(14)	110.3(3)
10	C(8)—C(9)—C(11)	111.7(3)	C(5)—C(10)—C(1)	110.8(2)
	C(9)—C(11)—C(12)	112.5(3)	C(13)—C(12)—C(11)	112.2(3)
	C(12)—C(13)—C(18)	110.6(3)	C(12)—C(13)—C(14)	107.8(3)
	C(18)—C(13)—C(14)	110.8(3)	C(12)—C(13)—C(17)	117.3(3)
	C(18)—C(13)—C(17)	109.7(3)	C(14)—C(13)—C(17)	100.1(2)
15	C(15)—C(14)—C(8)	121.0(3)	C(15)—C(14)—C(13)	104.6(3)
	C(8)—C(14)—C(13)	113.8(2)	C(14)—C(15)—C(16)	104.2(3)
	C(15)—C(16)—C(17)	107.4(3)	C(20)—C(17)—C(16)	111.3(3)
	C(20)—C(17)—C(13)	121.2(3)	C(16)—C(17)—C(13)	103.0(2)
20	C(22)—C(20)—C(17)	115.0(3)	C(22)—C(20)—C(21)	109.6(3)
	C(17)—C(20)—C(21)	110.8(3)	C(23)—C(22)—C(20)	115.7(4)

TABLE 4

Anisotropic displacement parameters (Å² × 10³) for 2-MbisP. The anisotropic displacement factor exponent takes the form: -2π²[h² α*² U₁₁ + ... + 2 h k α* b* U₁₂]

	U ₁₁	U ₂₂	U ₃₃	U ₂₃	U ₁₃	U ₁₂	
30	O(1)	36(1)	59(1)	92(2)	-7(1)	-3(1)	-5(1)
	O(3)	42(1)	48(1)	145(2)	3(1)	-17(2)	2(1)
	C(1)	34(2)	45(2)	70(2)	-5(2)	-11(2)	-1(1)
	C(2)	39(2)	51(2)	83(2)	-21(2)	-14(2)	6(1)
	C(3)	55(2)	52(2)	75(2)	-19(2)	-7(2)	8(2)
	C(4)	60(2)	56(2)	61(2)	-2(2)	2(2)	0(2)
35	C(5)	52(2)	34(1)	58(2)	3(1)	-5(2)	1(1)
	C(6)	51(2)	45(2)	61(2)	2(1)	5(2)	-9(2)
	C(7)	56(2)	42(2)	62(2)	2(1)	6(2)	-8(2)
	C(8)	47(2)	44(2)	58(2)	3(1)	-1(2)	-10(2)
	C(9)	48(2)	75(2)	73(2)	-15(2)	3(2)	-12(2)
	C(10)	44(2)	41(2)	66(2)	-11(1)	-6(2)	4(1)
40	C(11)	59(2)	78(2)	72(2)	-12(2)	11(2)	-32(2)
	C(12)	59(2)	61(2)	63(2)	-5(2)	1(2)	-22(2)
	C(13)	43(2)	51(2)	47(2)	5(1)	-6(2)	-7(2)
	C(14)	45(2)	52(2)	53(2)	4(1)	-4(2)	-8(2)
	C(15)	53(2)	64(2)	69(2)	1(2)	9(2)	-19(2)
	C(16)	48(2)	71(2)	58(2)	3(2)	4(2)	0(2)
45	C(17)	49(2)	53(2)	50(2)	1(1)	-7(2)	-1(2)
	C(18)	71(3)	66(2)	57(2)	10(2)	-11(2)	-5(2)
	C(19)	53(3)	122(4)	117(3)	-68(3)	-9(3)	-8(2)
	C(20)	73(3)	59(2)	64(2)	-5(2)	-2(2)	8(2)
	C(21)	117(4)	82(3)	76(2)	-16(2)	20(3)	9(3)
	C(22)	104(4)	63(2)	99(3)	-12(2)	18(3)	-6(2)
50	C(23)	107(4)	101(4)	140(5)	-39(3)	-4(4)	-25(3)

TABLE 5

Hydrogen coordinates (×10⁴) and isotropic displacement parameters (Å² × 10³) for 2-MbisP.

	x	y	z	U(eq)	
60	H(1O)	2030	2820	5188	75
	H(3O)	9248	3381	4573	94
	H(1)	4744	3489	5201	60
	H(3)	6972	3366	3580	73
	H(4A)	8510	4644	4021	71
	H(4B)	6418	4808	3682	71
	H(6)	8807	5506	4904	63
	H(7)	5716	5796	5840	64
65	H(9A)	10678	6347	5395	78
	H(9B)	11311	6330	6144	78

TABLE 5-continued

Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for 2-MbisP.				
	x	y	z	U(eq)
H(10A)	3261	4960	4528	61
H(10B)	3817	4933	5284	61
H(11A)	9979	7806	5444	84
H(11B)	12157	7673	5728	84
H(12A)	10992	7820	6807	74
H(12B)	10264	8624	6403	74
H(14)	8690	6614	6986	60
H(15A)	5627	6117	7210	74
H(15B)	4592	6682	6655	74
H(16A)	5482	7201	7925	71
H(16B)	4186	7714	7405	71
H(17)	8337	7857	7652	61
H(18A)	6780	8804	6077	97

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TABLE 5-continued

Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for 2-MbisP.				
	x	y	z	U(eq)
H(18B)	5109	8148	6295	97
H(18C)	6593	7911	5719	97
H(19A)	3904	2689	3412	117
H(19B)	2083	2780	3935	117
H(20)	5547	9133	7268	78
H(21A)	7037	8779	8551	138
H(21B)	4785	8697	8323	138
H(21C)	5783	9609	8388	138
H(22A)	8703	9627	6950	106
H(22B)	7832	10203	7518	106
H(23A)	9974	9502	8273	174
H(23B)	11194	9978	7717	174
H(23C)	10902	8976	7684	174

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TABLE 6

Torsion angles [$^\circ$] for 2-MbisP.			
O(1)—C(1)—C(2)—C(19)	0.5(5)	C(10)—C(1)—C(2)—C(19)	-122.6(4)
O(1)—C(1)—C(2)—C(3)	179.9(3)	C(10)—C(1)—C(2)—C(3)	56.8(3)
C(19)—C(2)—C(3)—O(3)	-111.3(4)	C(1)—C(2)—C(3)—O(3)	69.2(3)
C(19)—C(2)—C(3)—C(4)	125.8(4)	C(1)—C(2)—C(3)—C(4)	-53.7(4)
O(3)—C(3)—C(4)—C(5)	-71.4(4)	C(2)—C(3)—C(4)—C(5)	48.9(4)
C(3)—C(4)—C(5)—C(6)	128.0(3)	C(3)—C(4)—C(5)—C(10)	-50.1(4)
C(10)—C(5)—C(6)—C(7)	-2.4(5)	C(4)—C(5)—C(6)—C(7)	179.7(3)
C(5)—C(6)—C(7)—C(8)	-178.8(3)	C(6)—C(7)—C(8)—C(9)	1.4(5)
C(6)—C(7)—C(8)—C(14)	-179.6(3)	C(7)—C(8)—C(9)—C(11)	125.7(3)
C(14)—C(8)—C(9)—C(11)	-53.4(4)	C(6)—C(5)—C(10)—C(1)	-125.0(3)
C(4)—C(5)—C(10)—C(1)	53.1(3)	O(1)—C(1)—C(10)—C(5)	179.9(2)
C(2)—C(1)—C(10)—C(5)	-54.7(3)	C(8)—C(9)—C(11)—C(12)	53.0(4)
C(9)—C(11)—C(12)—C(13)	-54.6(4)	C(11)—C(12)—C(13)—C(18)	-66.8(4)
C(11)—C(12)—C(13)—C(14)	54.5(4)	C(11)—C(12)—C(13)—C(17)	166.4(3)
C(7)—C(8)—C(14)—C(15)	3.9(5)	C(9)—C(8)—C(14)—C(15)	-177.0(3)
C(7)—C(8)—C(14)—C(13)	-122.0(3)	C(9)—C(8)—C(14)—C(13)	57.2(4)
C(12)—C(13)—C(14)—C(15)	168.8(3)	C(18)—C(13)—C(14)—C(15)	-70.1(3)
C(17)—C(13)—C(14)—C(15)	45.6(3)	C(12)—C(13)—C(14)—C(8)	-57.1(3)
C(18)—C(13)—C(14)—C(8)	64.1(4)	C(17)—C(13)—C(14)—C(8)	179.7(3)
C(8)—C(14)—C(15)—C(16)	-163.1(3)	C(13)—C(14)—C(15)—C(16)	-33.1(3)
C(14)—C(15)—C(16)—C(17)	7.4(4)	C(15)—C(16)—C(17)—C(20)	151.9(3)
C(15)—C(16)—C(17)—C(13)	20.6(3)	C(12)—C(13)—C(17)—C(20)	79.2(4)
C(18)—C(13)—C(17)—C(20)	-48.2(4)	C(14)—C(13)—C(17)—C(20)	-164.7(3)
C(12)—C(13)—C(17)—C(16)	-155.7(3)	C(18)—C(13)—C(17)—C(16)	76.9(3)
C(14)—C(13)—C(17)—C(16)	-39.5(3)	C(16)—C(17)—C(20)—C(22)	-178.0(3)
C(13)—C(17)—C(20)—C(22)	-56.8(5)	C(16)—C(17)—C(20)—C(21)	56.9(4)
C(13)—C(17)—C(20)—C(21)	178.1(3)	C(17)—C(20)—C(22)—C(23)	-62.9(5)
C(21)—C(20)—C(22)—C(23)	62.8(5)		

TABLE 7

Observed and calculated structure factors for 2-Mbis-P

h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s
4	0	0	51	50	3	1	15	0	43	53	8	-4	14	1	298	328	8	-4	14	1	38	38	5
5	0	0	49	54	4	2	15	0	82	87	6	-3	14	1	326	324	6	-3	14	1	54	53	3
1	1	0	108	125	1	3	15	0	53	49	3	-2	14	1	62	70	2	-1	14	1	95	101	3
2	1	0	833	823	10	0	16	0	28	24	8	3	6	1	76	73	2	-1	14	1	78	78	2
4	1	0	448	435	4	1	16	0	25	18	5	4	6	1	68	62	1	0	14	1	73	68	1
5	1	0	62	62	1	2	16	0	17	12	8	5	6	1	125	129	2	1	14	1	78	79	3
6	1	0	65	72	2	1	17	0	54	37	10	6	6	1	89	91	2	2	14	1	97	101	2
0	2	0	650	820	8	1	0	1	304	297	5	-5	7	1	152	158	5	3	14	1	44	53	5
1	2	0	510	519	6	2	0	1	85	79	1	-4	7	1	168	165	1	4	14	1	44	34	5
2	2	0	88	88	2	4	0	1	115	107	1	-3	7	1	138	136	2	-3	15	1	33	5	5
4	2	0	217	213	2	5	0	1	111	115	2	-2	7	1	101	94	2	-2	15	1	46	46	6
5	2	0	120	117	1	6	0	1	36	28	4	-1	7	1	219	219	3	-1	15	1	25	30	6
6	2	0	36	42	5	-6	1	1	56	55	3	0	7	1	223	236	3	0	15	1	38	30	2
7	2	0	37	49	7	-5	1	1	107	120	2	1	7	1	213	220	3	1	15	1	38	30	5
1	3	0	352	375	3	-4	1	1	162	152	4	2	7	1	91	93	2	2	15	1	34	46	6
3	3	0	39	35	3	-2	1	1	338	350	2	3	7	1	136	137	1	3	15	1	27	4	5
3	3	0	58	55	2	-1	1	1	1141	1205	14	4	7	1	174	165	1	-2	16	1	24	27	6
4	3	0	84	80	1	1	1	1	1134	1206	11	5	7	1	160	158	2	-1	16	1	27	13	6
5	3	0	43	42	8	4	1	1	163	152	1	6	7	1	33	17	24	0	16	1	18	4	17
6	3	0	54	51	5	5	1	1	109	120	2	-6	8	1	61	46	10	1	16	1	33	13	8
7	3	0	34	7	7	6	1	1	56	55	11	-5	8	1	65	60	11	2	16	1	20	28	6
0	4	0	114	118	2	7	1	1	19	16	18	-4	8	1	52	51	2	0	17	1	24	32	5
1	4	0	17	3	3	-7	2	1	38	15	7	-3	8	1	150	145	1	1	17	1	17	3	17
2	4	0	88	82	1	-6	2	1	41	44	4	-2	8	1	86	88	1	0	0	2	393	405	5
4	4	0	87	87	1	-5	2	1	28	16	5	-1	8	1	68	69	5	1	0	2	332	336	8
5	4	0	19	15	18	-4	2	1	256	252	5	0	8	1	378	391	3	2	0	2	412	409	8
6	4	0	99	104	5	-3	2	1	108	98	2	1	8	1	69	68	2	4	0	2	137	137	3
7	4	0	66	56	4	-2	2	1	98	100	2	2	8	1	89	87	2	-6	1	3	34	44	5
1	5	0	17	4	7	-1	2	1	465	481	4	3	8	1	152	145	1	-5	1	2	114	119	4
2	5	0	72	64	1	0	2	1	15	2	3	4	8	1	54	51	6	-4	1	2	52	49	2
3	5	0	108	106	6	1	2	1	478	482	4	5	8	1	67	59	4	-2	1	2	174	178	1
4	5	0	86	82	1	2	2	1	93	99	2	6	8	1	46	48	4	-1	1	2	599	599	5
5	5	0	34	29	3	4	2	1	257	252	2	-6	9	1	37	28	12	0	1	2	166	164	2
6	5	0	29	22	6	5	2	1	17	16	8	-5	9	1	84	71	3	2	1	2	278	178	2
7	5	0	57	41	6	6	2	1	41	44	6	-4	9	1	105	103	1	4	1	2	50	49	2
0	6	0	645	708	8	7	2	1	17	14	16	-3	9	1	156	151	1	5	1	2	119	118	2
1	6	0	360	437	50	-7	3	1	41	49	6	-2	9	1	54	46	1	6	1	2	47	44	3
2	6	0	101	86	21	-6	3	1	171	175	4	-1	9	1	194	201	9	7	1	2	59	55	5
3	6	0	36	39	3	-5	3	1	239	231	3	1	9	1	227	223	2	-7	2	2	39	34	6
4	6	0	19	17	4	-4	3	1	457	454	6	2	9	1	190	201	7	-6	2	2	106	111	2
5	6	0	80	76	4	-3	3	1	457	454	6	2	9	1	50	46	3	-5	2	2	211	211	3
6	6	0	48	44	4	-2	3	1	306	308	4	3	9	1	157	151	1	-4	2	2	74	77	1
1	7	0	31	28	3	-1	3	1	375	403	4	4	9	1	114	103	1	-3	2	2	385	373	5
2	7	0	110	128	25	0	3	1	358	411	19	5	9	1	74	71	5	-2	2	2	96	91	1
3	7	0	72	70	1	1	3	1	394	402	3	6	9	1	36	28	6	-1	2	2	614	632	19

TABLE 7-continued

Observed and calculated structure factors for 2-Mbis-P

h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s
4	7	0	79	78	2	2	3	1	304	307	4	-5	10	1	54	57	4	0	2	2	371	373	7
5	6	0	171	167	4	3	3	1	472	455	6	-4	10	1	107	110	3	1	2	2	623	632	8
6	7	0	49	48	4	4	3	1	240	231	2	-3	10	1	168	166	1	2	2	2	105	92	2
0	8	0	134	137	2	5	3	1	169	176	3	-2	10	1	79	79	3	4	2	2	80	78	1
1	8	0	226	226	8	6	3	1	44	50	5	-1	10	1	139	149	2	5	2	2	211	211	6
2	8	0	211	196	3	7	3	1	54	49	5	0	10	1	127	138	1	6	2	2	116	111	3
3	8	0	66	64	1	-7	4	1	46	41	5	1	10	1	145	148	1	7	2	2	34	35	7
4	8	0	315	302	2	-6	4	1	81	84	2	2	10	1	74	79	1	-7	3	2	69	71	4
5	8	0	67	53	3	-5	4	1	36	26	20	3	10	1	168	165	1	-6	3	2	75	81	4
6	8	0	38	28	6	-4	4	1	171	165	1	4	10	1	107	110	2	-5	3	2	112	104	1
1	9	0	247	260	7	-3	4	1	189	193	3	5	10	1	62	57	4	-4	3	2	147	144	3
2	9	0	243	244	2	-2	4	1	199	199	3	-5	11	1	22	6	13	-3	3	2	190	177	3
3	9	0	193	186	2	-1	4	1	346	367	4	-4	11	1	77	79	2	-2	3	2	105	107	2
4	9	0	405	398	3	0	4	1	450	520	13	-3	11	1	31	50	1	-1	3	2	395	396	14
5	9	0	8	2	8	2	8	1	356	368	13	-2	11	1	70	71	1	0	3	2	252	272	12
6	9	0	91	75	7	2	4	1	195	200	2	-1	11	1	92	93	2	1	3	2	391	396	5
0	10	0	119	124	2	3	4	1	182	194	3	0	11	1	98	100	2	2	3	2	108	107	2
1	10	0	54	49	1	4	4	1	176	166	3	1	11	1	94	93	2	4	3	2	146	144	1
2	10	0	127	124	1	5	4	1	37	26	4	2	11	1	58	71	1	5	3	2	108	103	4
3	10	0	116	115	1	6	4	1	83	84	3	3	11	1	52	50	2	6	3	2	89	81	3
4	10	0	26	0	5	7	4	1	37	41	6	4	11	1	79	79	2	7	3	2	75	70	3
5	10	0	63	63	4	-7	5	1	58	45	5	5	11	1	6	6	5	-6	4	2	95	101	2
1	11	0	77	75	2	-6	5	1	26	18	16	-5	12	1	49	46	6	-5	4	2	75	79	4
2	11	0	46	45	3	-5	5	1	79	76	2	-4	12	1	40	35	4	-4	4	2	172	175	3
3	11	0	15	15	5	-4	5	1	44	46	1	-3	12	1	52	55	2	-3	4	2	241	230	3
5	11	0	142	441	2	-3	5	1	136	137	2	-2	12	1	162	161	1	-2	4	2	431	439	5
0	12	0	46	35	14	-2	5	1	337	336	4	-1	12	1	159	150	4	-1	4	2	58	62	2
1	12	0	58	55	3	-1	5	1	171	174	2	0	12	1	111	117	2	0	4	2	242	268	20
2	12	0	67	69	1	0	5	1	274	294	2	1	12	1	163	151	3	1	4	2	69	63	1
3	12	0	145	145	1	1	5	1	174	174	2	2	12	1	158	161	1	2	4	2	438	438	7
4	12	0	43	40	4	2	5	1	343	336	4	3	12	1	56	55	2	3	4	2	247	229	3
5	12	0	32	24	4	3	5	1	126	137	1	4	12	1	32	35	10	4	4	2	177	175	2
1	13	0	26	9	7	5	5	1	92	76	3	-4	13	1	41	23	9	6	4	2	106	101	3
2	13	0	161	164	1	6	5	1	24	18	14	-3	13	1	67	71	2	7	4	2	50	40	5
3	13	0	137	141	1	7	5	1	88	45	4	-2	13	1	45	47	2	-6	5	2	85	79	4
4	13	0	43	42	9	-6	6	1	125	128	1	-1	13	1	23	25	3	-4	5	2	188	188	1
0	14	0	79	84	2	-5	6	1	61	61	1	1	13	1	54	38	5	-3	5	2	162	167	3
1	14	0	40	32	2	-4	6	1	77	73	1	2	13	1	50	47	4	-2	5	2	181	176	3
2	14	0	32	26	4	-3	6	1	73	71	2	3	13	1	70	71	2	0	5	2	151	146	2
3	14	0	87	89	2	-2	6	1	327	325	4	4	13	1	41	23	7	0	5	2	901	990	15
4	14	0	49	43	4	-1	6	1	305	297	3	0	11	3	72	67	1	1	3	4	316	326	2
1	12	2	38	27	9	1	4	3	115	120	2	2	11	3	65	63	2	4	4	4	11	8	10
2	12	2	43	48	2	2	4	3	86	85	2	2	11	3	141	137	2	5	3	4	38	44	3
3	12	2	41	48	2	4	4	3	139	136	3	3	11	3	159	153	1	6	3	4	81	78	3
4	12	2	66	59	3	5	4	3												4			

TABLE 7-continued

Observed and calculated structure factors for 2-Mbis-P

h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s
5	12	2	33	63	9	6	4	3	72	68	4	4	11	3	130	130	2	7	3	4	42	41	5
4	13	2	87	34	8	7	4	3	48	42	5	5	11	3	96	92	2	-6	4	4	84	94	3
3	13	2	34	90	2	-6	5	3	52	51	11	-5	12	3	38	48	8	-5	4	4	28	32	4
2	13	2	36	20	5	-5	5	3	172	171	2	-4	12	3	29	25	5	-4	4	4	142	142	1
1	13	2	116	120	2	-4	5	3	76	72	2	-3	12	3	135	135	4	-3	4	4	235	222	3
0	13	2	66	62	2	-3	5	3	21	12	7	-2	12	3	80	84	2	-2	4	4	186	180	2
1	13	2	117	120	3	-2	5	3	223	216	2	-1	12	3	181	175	3	-1	4	4	400	390	8
2	13	2	30	20	7	-1	5	3	186	196	2	0	12	3	45	49	2	0	4	4	327	325	3
3	13	2	89	90	2	0	5	3	733	738	46	1	12	3	176	174	3	1	4	4	392	390	9
4	13	2	38	34	5	1	5	3	197	196	2	2	12	3	81	84	3	2	4	4	186	180	6
4	14	2	60	57	8	2	5	3	225	217	3	3	12	3	133	136	1	3	4	4	208	222	3
3	14	2	41	43	3	3	5	3	15	12	10	4	12	3	15	24	14	4	4	4	153	142	2
2	14	2	56	61	2	4	5	3	79	73	2	5	12	3	50	48	3	5	4	4	31	32	4
1	14	2	34	25	3	5	5	3	176	171	4	-4	13	3	25	20	13	6	4	4	90	94	3
0	14	2	92	92	1	6	5	3	54	51	5	-3	13	3	64	62	2	7	4	4	25	6	10
1	14	2	22	25	6	-6	6	3	52	51	7	-2	13	3	38	36	2	-6	5	4	84	99	4
2	14	2	61	61	4	-4	6	3	93	93	3	0	13	3	49	48	2	-5	5	4	133	138	5
3	14	2	41	42	4	-4	6	3	120	114	2	0	13	3	34	24	5	-4	5	4	57	59	2
4	14	2	53	57	5	-3	6	3	62	60	2	1	13	3	44	48	2	-3	5	4	147	150	2
3	15	2	36	36	4	-2	6	3	238	243	3	2	13	3	48	36	12	-2	5	4	177	182	3
2	15	2	47	44	4	-1	6	3	88	86	2	3	13	3	63	62	3	-1	5	4	258	237	2
1	15	2	21	10	7	0	6	3	136	146	1	4	13	3	38	20	8	0	5	4	339	335	20
0	15	2	33	36	3	1	6	3	92	84	1	-4	14	3	37	38	8	1	5	4	247	237	1
1	15	2	25	10	5	2	6	3	233	242	3	-3	14	3	96	96	2	2	5	4	165	181	1
2	15	2	40	44	4	3	6	3	61	61	2	-2	14	3	94	106	2	3	5	4	132	151	2
3	15	2	34	37	4	4	6	3	125	114	2	-1	14	3	31	18	4	4	5	4	60	59	2
4	15	2	24	16	5	5	6	3	97	93	2	0	14	3	27	29	4	5	5	4	132	138	2
5	15	2	27	29	6	6	6	3	58	51	4	1	14	3	24	19	5	6	5	4	94	99	3
6	15	2	37	17	4	4	6	3	32	23	9	2	14	3	101	106	3	-6	6	4	66	73	11
7	15	2	29	29	3	-5	7	3	33	57	2	3	14	3	99	96	5	-5	6	4	59	58	3
8	15	2	23	16	6	-4	7	3	232	230	2	4	14	3	37	38	4	-4	6	4	77	79	2
9	15	2	19	14	8	-3	7	3	59	56	2	-3	15	3	35	35	9	-3	6	4	67	64	1
10	15	2	37	28	6	-2	7	3	171	151	3	-2	15	3	36	33	3	-2	6	4	173	169	3
11	15	2	986	1006	11	-1	7	3	228	222	3	-1	15	3	32	21	6	-1	6	4	330	348	4
0	16	2	115	121	2	0	7	3	379	403	3	0	15	3	39	34	3	0	6	4	55	58	4
1	16	2	324	311	5	1	7	3	230	222	6	1	15	3	36	21	7	1	6	4	350	349	7
2	16	2	56	51	14	2	7	3	167	180	3	2	15	3	31	33	6	2	6	4	157	169	2
3	16	2	73	67	3	3	7	3	57	56	1	3	15	3	40	35	4	3	6	4	48	64	1
4	16	2	253	250	9	4	7	3	231	230	7	-2	16	3	34	35	2	4	6	4	82	79	6
5	16	2	176	170	3	5	7	3	59	57	3	-1	16	3	28	7	5	5	6	4	58	58	2
6	16	2	392	392	7	6	7	3	38	23	16	0	16	3	52	35	2	6	6	4	77	73	4
7	16	2	207	207	2	-6	8	3	32	28	10	1	16	3	29	7	6	-6	7	4	55	36	6
8	16	2	101	95	2	-5	8	3	31	15	4	2	16	3	42	36	6	-5	7	4	101	105	3
9	16	2	1115	1114	19	-4	8	3	73	73	2	0	17	3	68	76	3	-4	7	4	57	57	2
10	16	2	97	96	2	-3	8	3	293	191	2	0	0	4	372	431	7	-3	7	4	35	36	1
11	16	2	205	206	2	-2	8	3	189	190	4	1	0	4	549	576	4	-2	7	4	147	139	3

TABLE 7-continued

Observed and calculated structure factors for 2-Mbis-P.

h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s
4	1	3	176	169	2	-1	8	3	310	324	4	2	0	4	64	65	1	-1	7	4	302	287	4
5	1	3	252	249	5	0	8	3	234	243	11	4	0	4	38	42	23	0	7	4	175	175	14
6	1	3	69	66	2	1	8	3	308	325	7	5	0	4	99	100	3	1	7	4	299	287	4
7	1	3	42	18	6	2	8	3	185	191	3	6	0	4	116	121	7	2	7	4	142	175	2
6	2	3	77	80	2	3	8	3	188	191	2	-6	1	4	66	64	5	3	7	4	39	287	2
5	2	3	59	70	2	4	8	3	79	78	2	-5	1	4	100	102	2	4	7	4	58	139	2
4	2	3	87	91	3	5	8	3	27	15	22	-4	1	4	90	77	2	5	7	4	108	36	2
3	2	3	223	231	3	6	8	3	25	28	8	-3	1	4	141	148	1	6	7	4	69	57	5
2	2	3	377	364	3	-6	9	3	59	52	6	-2	1	4	112	111	2	-6	8	4	45	106	4
1	2	3	725	744	16	-5	9	3	82	80	4	-1	1	4	489	481	4	-5	8	4	176	36	2
0	2	3	1539	1543	19	-4	9	3	261	247	2	0	1	4	731	731	10	-4	8	4	86	36	1
1	1	2	723	744	8	-3	9	3	116	117	3	1	1	4	473	481	6	-3	8	4	207	177	6
2	2	3	359	364	4	-2	9	3	72	67	2	2	1	4	108	111	2	-2	8	4	119	93	3
4	2	3	83	91	1	0	9	3	40	31	3	4	1	4	90	77	4	-1	8	4	231	208	6
5	2	3	63	70	1	1	9	3	52	55	2	5	1	4	102	103	5	0	8	4	180	121	10
6	2	3	86	80	3	1	9	3	32	31	1	6	1	4	74	64	4	1	8	4	236	230	7
7	2	3	25	20	11	3	9	3	69	67	1	7	1	4	38	45	7	2	8	4	124	122	2
6	3	3	79	84	4	3	9	3	116	117	1	-6	2	4	30	20	9	3	8	4	211	208	4
5	3	3	127	121	2	4	9	3	256	247	4	-5	2	4	203	150	3	4	8	4	91	93	2
4	3	3	217	214	8	5	9	3	83	80	5	-4	2	4	136	129	2	5	8	4	181	177	2
3	3	3	87	77	2	6	9	3	66	52	9	-3	2	4	329	324	3	6	8	4	46	36	5
2	3	3	217	219	2	-5	10	3	65	65	4	-2	2	4	236	241	2	-6	9	4	48	41	7
1	3	3	676	688	22	-4	10	3	146	144	1	-1	2	4	471	497	15	-5	9	4	51	48	3
0	3	3	461	472	10	-3	10	3	37	35	2	0	2	4	952	922	20	-4	9	4	107	104	3
1	3	3	675	687	5	-2	10	3	85	74	3	1	2	4	483	497	8	-3	9	4	74	74	1
2	3	3	228	219	3	-1	10	3	251	228	2	2	2	4	232	242	3	-2	9	4	101	98	2
4	3	3	220	213	2	0	10	3	63	41	1	4	2	4	125	129	1	-1	9	4	72	69	1
5	3	3	130	120	7	1	10	3	237	228	8	5	2	4	203	191	3	0	9	4	176	174	4
6	3	3	81	84	3	2	10	3	83	74	1	6	2	4	27	20	11	1	9	4	66	69	1
7	3	3	54	45	5	3	10	3	34	35	2	7	2	4	38	9	7	2	9	4	102	97	4
6	4	3	69	68	3	4	10	3	147	144	3	-6	3	4	82	78	3	3	9	4	78	73	2
5	4	3	135	136	2	5	10	3	147	144	3	-5	3	4	44	44	7	4	9	4	104	104	2
4	4	3	79	85	1	-5	11	3	96	91	4	-4	3	4	15	9	9	5	9	4	53	48	5
3	4	3	109	103	2	-4	11	3	130	130	3	-3	3	4	275	287	4	6	9	4	36	41	6
2	4	3	119	121	2	-3	11	3	157	153	4	-2	3	4	262	259	7	-5	10	4	59	60	8
1	4	3	300	297	9	-2	11	3	148	137	2	-1	3	4	315	325	15	-4	10	4	74	66	3
0	4	3	261	268	3	-1	11	3	58	63	1	0	3	4	241	241	4	-3	10	4	104	104	2
5	2	5	124	119	2	-3	10	5	73	75	1	7	2	6	35	28	8	4	10	6	59	52	4
6	2	5	66	56	4	-2	10	5	32	29	2	-4	3	6	147	143	2	5	10	6	92	97	2
7	2	5	38	37	5	-1	10	5	109	112	2	-3	3	6	127	134	2	-5	11	6	34	22	7
5	3	5	78	76	3	0	10	5	20	21	5	-2	3	6	361	357	3	-4	11	6	129	126	2
4	3	5	105	98	2	1	10	5	111	112	2	-1	3	6	287	275	2	-3	11	6	30	29	6
3	3	5	226	216	2	2	10	5	39	29	8	0	3	6	283	262	12	-2	11	6	113	115	1
2	3	5	206	194	4	3	10	5	70	75	1	1	3	6	273	276	2	-1	11	6	170	176	5
1	3	5	158	136	2	4	10	5	43	41	4	2	3	6	356	357	4	0	11	6	23	20	2
0	3	5	353	335	5	5	10	5	108	110	4	4	4	6	145	143	3	1	11	6	170	176	2

TABLE 7-continued

Observed and calculated structure factors for 2-Mbis-P

h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s
1	3	5	159	136	3	-5	11	5	33	27	6	5	3	6	201	198	3	2	11	6	111	116	2
4	3	5	99	97	3	-4	11	5	90	84	3	6	3	6	61	57	5	4	11	6	31	29	16
5	3	5	80	76	3	-3	11	5	29	27	2	3	4	6	86	87	2	4	11	6	124	125	3
6	3	5	60	51	5	-2	11	5	104	99	1	-2	4	6	113	101	1	5	11	6	23	22	7
7	3	5	32	6	8	-1	11	5	97	106	2	-1	4	6	539	523	13	-4	12	6	38	33	6
4	4	5	20	21	19	0	11	5	28	2	8	0	4	6	65	59	1	-3	12	6	129	128	4
3	4	5	203	196	3	1	11	5	102	106	2	1	4	6	530	523	6	-2	12	6	100	93	3
2	4	5	248	242	5	2	11	5	100	100	1	2	4	6	102	101	2	0	12	6	91	89	3
1	4	5	73	71	2	3	11	5	39	27	6	3	4	6	87	87	2	0	12	6	111	112	1
0	4	5	158	171	5	4	11	5	83	84	3	4	4	6	103	99	3	1	12	6	90	89	2
3	4	5	180	196	3	-4	12	5	52	55	4	6	4	6	32	54	2	6	12	6	94	94	2
4	4	5	8	20	7	-3	12	5	53	44	4	-4	5	6	62	62	2	4	12	6	30	32	7
5	4	5	58	63	2	-1	12	5	86	81	1	-3	5	6	27	8	3	13	6	28	16	12	
6	4	5	126	122	3	-2	12	5	39	41	2	-2	5	6	166	164	2	-3	13	6	114	124	5
5	5	5	46	42	5	0	12	5	31	3	7	-1	5	6	301	282	3	-2	13	6	104	105	2
4	5	5	26	23	3	1	12	5	43	41	2	0	5	6	109	98	1	-1	13	6	117	121	4
3	5	5	50	47	1	2	12	5	82	81	1	1	5	6	291	282	4	0	13	6	26	25	4
2	5	5	428	416	4	3	12	5	50	44	3	3	5	6	10	8	9	1	13	6	117	121	2
1	5	5	126	136	2	4	12	5	56	54	3	4	5	6	68	62	1	2	13	6	103	105	1
0	5	5	275	251	4	5	12	5	41	29	5	5	5	6	34	32	8	3	13	6	123	124	3
1	5	5	120	136	4	-4	13	5	31	22	10	6	5	6	83	73	4	4	13	6	27	14	8
3	5	5	50	47	1	-3	13	5	51	47	5	-5	6	6	68	64	5	-2	14	6	21	28	6
4	5	5	26	23	4	-2	13	5	72	71	4	-4	6	6	52	48	2	-2	14	6	56	46	5
5	5	5	46	42	4	-1	13	5	50	51	6	-3	6	6	112	107	1	-1	14	6	10	18	10
6	5	5	100	92	3	0	13	5	33	33	5	-2	6	6	68	63	2	0	14	6	53	49	2
5	6	5	65	62	5	1	13	5	53	51	2	-1	6	6	172	175	4	1	14	6	26	18	22
4	6	5	56	57	2	2	13	5	71	72	3	0	6	6	260	239	25	2	14	6	47	47	7
3	6	5	95	97	1	3	13	5	50	47	3	1	6	6	168	175	2	3	14	6	31	28	5
2	6	5	173	162	4	4	13	5	33	22	5	2	6	6	60	63	2	-2	15	6	37	40	6
1	6	5	224	223	3	-3	14	5	38	39	4	3	6	6	113	147	1	-1	15	6	42	42	12
0	6	5	15	6	14	-2	14	5	88	91	2	4	6	6	49	48	2	0	15	6	61	68	6
1	6	5	223	222	3	-1	14	5	43	41	5	5	6	6	67	63	4	1	15	6	42	43	4
2	6	5	165	163	2	0	14	5	39	61	2	6	6	6	62	68	4	2	15	6	42	40	7
3	6	5	100	98	2	1	14	5	41	41	2	-5	7	6	124	120	5	3	15	6	56	57	5
4	6	5	64	56	1	2	14	5	92	91	4	-4	7	6	57	58	1	-1	16	6	51	52	6
5	6	5	61	62	4	3	14	5	35	39	9	-3	7	6	188	193	3	0	16	6	12	1	12
6	7	5	76	68	4	-3	15	5	36	31	10	-2	7	6	53	55	1	1	16	6	52	52	3
6	7	5	24	64	23	-2	15	5	74	80	2	-1	7	6	134	123	2	1	0	7	35	42	6
5	7	5	83	89	4	-1	15	5	45	51	10	14	0	7	171	186	14	2	0	7	224	211	3
4	7	5	47	47	2	0	15	5	21	10	14	0	7	6	130	123	2	3	0	7	28	30	4
3	7	5	82	82	1	1	15	5	22	10	22	2	7	6	53	55	5	4	0	7	108	111	2
2	7	5	173	177	1	3	15	5	75	80	3	3	7	6	189	193	3	5	0	7	73	75	3
1	7	5	201	203	7	3	15	5	40	31	4	4	7	6	53	57	2	6	0	7	17	1	16
0	7	5	343	344	27	-2	16	5	26	10	9	5	7	6	118	120	2	-6	1	7	51	64	5
1	7	5	196	204	6	-1	16	5	64	63	3	6	7	6	43	35	6	-5	1	7	135	132	2

TABLE 7-continued

Observed and calculated structure factors for 2-Mbis-P.

h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	
2	7	5	171	177	1	0	16	5	35	67	61	9	4	1	7	180	176	2	9	7	22	6	21	
3	4	7	81	83	1	1	16	5	59	63	47	2	-4	1	7	162	168	2	-4	9	7	63	62	2
3	4	7	5	46	46	2	16	5	25	10	35	1	-3	1	7	132	132	2	-3	9	7	85	90	1
5	7	5	82	89	2	0	0	6	402	451	65	2	-1	1	7	191	189	4	-2	9	7	121	119	1
6	7	5	63	64	4	1	0	6	69	75	1	0	1	7	250	250	2	-1	9	7	197	194	2	
6	8	5	43	20	7	2	0	6	27	29	4	18	13	12	1	7	180	189	2	0	7	178	178	1
5	8	5	69	59	7	3	0	6	85	80	2	1	11	4	7	128	132	2	1	9	7	188	194	6
4	8	5	117	117	1	5	0	6	160	166	3	2	8	6	7	163	168	2	2	9	7	117	118	2
3	8	5	133	128	2	6	0	6	112	109	3	3	8	6	7	190	176	3	3	9	7	97	90	3
2	8	5	63	62	2	-6	1	6	205	205	3	4	8	6	7	138	132	3	4	9	7	68	62	3
1	8	5	371	376	3	-5	1	6	89	87	2	5	8	6	7	63	63	4	-5	10	7	43	43	7
0	8	5	24	31	6	-4	1	6	132	125	2	6	8	6	7	93	90	3	-4	10	7	95	98	2
1	8	5	380	376	8	-3	1	6	215	217	3	-5	9	6	7	31	31	4	-3	10	7	160	131	1
2	8	5	61	62	2	-2	1	6	329	336	6	-4	9	6	7	61	70	2	-2	10	7	134	135	1
3	8	5	130	128	2	-1	1	6	174	173	1	-3	9	6	7	140	132	4	-1	10	7	127	125	3
4	8	5	120	117	4	0	1	6	155	165	1	-2	9	6	7	394	385	3	0	10	7	159	138	1
5	8	5	67	58	3	1	1	6	160	173	1	-1	9	6	7	111	114	1	1	10	7	125	125	1
6	8	5	33	20	7	2	1	6	333	336	4	0	9	6	7	378	385	6	2	10	7	135	135	1
5	9	5	28	33	8	3	1	6	207	218	3	1	9	6	7	37	30	3	3	10	7	154	151	3
4	9	5	96	98	1	4	1	6	127	124	2	2	9	6	7	87	91	3	4	10	7	93	98	3
3	9	5	94	92	1	4	1	6	91	84	2	2	9	6	7	67	76	4	5	10	7	31	43	6
2	9	5	138	136	2	6	1	6	207	205	3	4	9	6	7	203	197	3	-5	11	7	54	42	4
1	9	5	165	157	2	-5	2	6	199	193	3	5	9	6	7	104	106	2	-4	11	7	57	60	4
0	9	5	144	144	2	-4	2	6	77	79	2	-4	10	6	7	335	347	9	-3	11	7	57	49	3
1	9	5	158	157	4	-3	2	6	121	120	1	-4	10	6	7	387	366	3	-2	11	7	53	54	2
2	9	5	143	135	4	-2	2	6	295	299	7	-3	10	6	7	176	151	2	-1	11	7	53	46	4
3	9	5	91	91	2	-1	2	6	433	433	4	-2	10	6	7	375	367	5	0	11	7	42	39	2
4	9	5	99	98	4	0	2	6	271	248	2	-1	10	6	7	347	346	4	1	11	7	51	46	1
5	9	5	37	33	5	1	2	6	432	433	5	0	10	6	7	98	105	2	2	11	7	61	55	4
6	9	5	38	33	6	4	2	6	81	78	2	1	10	6	7	209	197	3	3	11	7	53	48	3
5	10	5	115	110	2	5	2	6	203	193	3	2	10	6	7	102	101	2	4	11	7	62	61	3
4	10	7	50	40	4	6	2	8	83	80	4	3	10	6	7	46	48	6	5	11	7	44	42	5
4	12	7	29	16	11	2	6	8	163	159	2	-1	14	8	9	89	87	2	2	10	177	185	3	
3	12	7	79	84	3	3	6	8	25	21	3	0	14	8	9	39	27	25	3	2	10	85	87	3
2	12	7	126	117	2	4	6	8	33	38	2	1	14	8	9	31	83	11	4	2	10	174	187	3
1	12	7	25	20	4	5	6	8	68	64	3	2	14	8	9	58	57	4	-4	3	10	182	198	3
0	12	7	308	305	6	6	6	8	104	100	3	3	14	8	9	52	47	2	-3	3	10	173	176	4
1	12	7	30	20	3	-4	6	8	57	59	2	-2	15	8	9	150	146	2	-2	3	10	121	111	1
2	12	7	111	117	2	-3	6	8	64	61	2	-1	15	8	9	74	76	2	-1	3	10	164	167	2
3	12	7	75	84	4	-2	6	8	152	140	2	0	15	8	9	94	99	1	0	3	10	121	121	1
4	12	7	18	16	17	-1	6	8	125	116	3	1	15	8	9	72	75	1	1	3	10	160	167	2
4	13	7	47	34	7	0	6	8	71	70	3	2	15	8	9	153	146	2	2	3	10	117	111	1
2	13	7	76	72	3	1	6	8	119	116	2	1	0	9	9	51	47	4	3	3	10	169	176	2
1	13	7	111	113	5	2	6	8	152	140	2	2	0	9	9	59	57	2	4	3	10	175	198	3
0	13	7	58	58	3	3	6	8	60	60	1	3	0	9	9	44	43	4	-4	4	10	71	64	3
1	13	7	109	113	1	4	6	8	59	59	1	-5	9	9	9	53	47	6	-3	4	10	56	55	1

TABLE 7-continued

Observed and calculated structure factors for 2-Mbis-P.

h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s
2	13	7	75	72	2	5	0	9	103	105	2	-4	9	9	144	146	2	-2	4	10	43	40	2
3	13	7	28	22	4	5	-5	9	121	119	3	-3	9	9	57	60	2	-1	4	10	99	88	1
4	13	7	34	34	6	-4	1	9	56	51	3	-2	9	9	29	21	7	0	4	10	328	328	8
3	14	7	29	28	20	-3	1	9	26	30	3	-1	9	9	38	29	6	1	4	10	91	68	1
1	14	7	36	30	10	-2	1	9	84	81	2	0	9	9	96	101	2	2	4	10	39	40	1
1	14	7	104	112	9	-1	1	9	150	152	7	1	9	9	43	29	8	3	4	10	61	55	3
0	14	7	36	35	3	0	1	9	179	191	6	2	9	9	36	20	4	4	4	10	63	64	3
1	14	7	108	112	2	1	1	9	147	152	1	3	9	9	62	60	3	5	4	10	61	77	3
2	14	7	28	30	6	2	1	9	80	81	2	4	9	9	149	146	3	-4	5	10	168	170	3
3	14	7	37	28	9	3	1	9	34	30	3	5	9	9	48	47	6	-3	5	10	73	73	2
2	15	7	56	59	3	4	7	8	47	51	3	-5	10	9	35	26	8	-2	5	10	121	118	2
1	15	7	63	69	9	5	1	9	128	119	3	-4	10	9	49	37	11	-1	5	10	150	146	1
0	15	7	94	88	2	6	1	9	51	41	5	-3	10	9	39	29	3	0	5	10	75	75	1
1	15	7	69	68	2	5	4	9	115	122	3	-2	10	9	112	110	1	1	5	10	147	146	2
2	15	7	56	59	3	-4	8	8	31	30	2	-1	10	9	246	241	2	2	5	10	115	118	4
1	16	7	48	45	3	-3	3	8	200	201	2	0	10	9	151	138	2	3	5	10	71	74	3
0	16	7	34	31	4	-2	1	9	284	274	3	1	10	9	248	241	2	4	5	10	147	170	3
0	0	8	458	523	44	-1	8	8	541	526	4	2	10	9	105	110	2	5	5	10	26	44	10
1	0	8	55	59	1	0	8	8	273	275	2	3	10	9	31	29	4	-4	6	10	69	66	4
2	0	8	226	227	3	1	2	9	192	201	3	4	10	9	39	37	4	-3	6	10	100	100	2
3	0	8	116	115	1	2	8	8	21	30	5	-4	11	9	35	44	6	-2	6	10	41	38	2
4	0	8	115	117	12	3	3	8	107	121	4	-3	11	9	106	104	2	-1	6	10	89	82	1
5	0	8	41	51	5	4	5	8	85	78	3	-2	11	9	176	175	2	0	6	10	198	187	2
6	0	8	85	79	4	5	4	8	100	95	3	-1	11	9	35	36	3	1	6	10	87	82	1
5	1	8	178	174	3	-5	9	8	103	105	3	0	11	9	29	18	3	2	6	10	37	38	2
4	1	8	91	83	2	-4	9	8	21	9	4	1	11	9	41	36	4	3	6	10	103	99	2
3	1	8	65	68	2	-3	9	8	233	217	2	2	11	9	170	175	10	4	6	10	66	67	3
2	1	8	233	239	7	-2	9	8	44	36	2	3	11	9	103	103	2	5	6	10	62	67	4
1	1	8	407	410	9	-1	9	8	220	217	3	4	11	9	48	44	3	-4	7	10	69	68	1
0	1	8	116	124	1	0	9	8	101	104	2	-4	12	9	65	61	3	-3	7	10	73	71	4
1	1	8	394	410	3	1	9	8	123	123	9	-3	12	9	84	83	5	-2	7	10	99	98	1
2	1	8	221	238	3	2	9	8	78	75	3	-2	12	9	63	60	3	-1	7	10	46	40	2
3	1	8	70	69	1	3	9	8	105	107	3	-1	12	9	140	134	1	0	7	10	48	47	2
4	1	8	85	82	6	4	9	8	62	70	3	0	12	9	43	26	4	1	7	10	41	40	2
5	1	8	181	173	3	5	9	8	110	111	1	1	12	9	140	134	1	2	7	10	103	98	4
6	1	8	82	74	4	-5	10	8	269	269	1	1	12	9	97	61	6	3	7	10	62	68	3
5	2	8	35	37	5	-4	10	8	132	129	2	3	12	9	85	83	4	4	7	10	66	71	3
4	2	8	42	40	4	-3	10	8	253	231	9	4	12	9	45	61	5	5	7	10	36	46	7
3	2	8	214	231	7	-2	10	8	126	128	1	-3	13	9	78	85	2	-4	8	10	64	64	3
2	2	8	120	121	2	-1	10	8	252	269	2	-2	13	9	67	63	2	-3	8	10	50	45	4
1	2	8	130	126	1	0	10	8	114	111	2	0	13	9	71	72	4	-2	8	10	103	109	1
0	2	8	172	160	2	1	10	8	133	136	2	4	13	9	78	76	2	-1	8	10	43	44	2
1	2	8	124	126	1	2	10	8	84	62	3	1	13	9	70	72	2	0	8	10	92	89	1
2	2	8	109	121	1	3	10	8	28	20	9	2	13	9	66	63	14	1	8	10	46	44	3
3	2	8	219	231	2	4	10	8	82	83	2	3	13	9	84	85	3	2	8	10	101	105	3
4	2	8	38	40	2	5	10	8	88	88	1	-2	14	9	29	16	10	3	8	10	46	45	4

TABLE 7-continued

Observed and calculated structure factors for 2-Mbis-P

h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s
5	2	8	46	37	4	-4	11	8	21	10	20	-2	5	9	85	82	2	-1	14	9	34	23	6
6	3	8	80	52	3	-3	11	8	123	123	3	-1	5	9	90	90	2	0	14	9	15	18	16
3	2	8	161	164	2	-2	11	8	105	105	1	0	5	9	66	72	2	1	14	9	16	23	15
2	3	8	80	80	3	-1	11	8	130	129	3	1	5	9	95	90	2	2	14	9	39	16	6
1	3	8	492	476	4	0	11	8	34	1	5	2	5	9	84	82	4	-2	15	9	54	51	4
0	3	8	44	42	2	1	11	8	128	129	1	3	5	9	88	88	2	-1	15	9	31	28	6
1	3	8	486	476	10	2	11	8	100	105	2	4	5	9	81	83	2	0	15	9	75	82	2
2	3	8	78	81	2	3	11	8	121	123	2	5	5	9	40	48	5	0	10	10	131	138	2
3	3	8	160	165	7	4	11	8	27	10	6	-4	6	9	109	107	2	1	0	10	112	115	2
4	3	8	84	85	1	-4	12	8	12	22	11	-3	6	9	31	36	2	2	0	10	4	19	4
5	3	8	64	59	3	-3	12	8	87	84	4	-2	6	9	121	121	1	3	0	10	176	186	3
6	3	8	66	60	5	-2	12	8	32	24	16	-1	6	9	105	110	1	4	0	10	105	123	2
3	4	8	133	122	1	-1	12	8	79	81	2	0	6	9	26	20	3	-4	1	10	167	184	2
2	4	8	182	179	5	0	12	8	242	234	3	1	6	9	104	110	2	-3	1	10	153	164	3
1	4	8	77	64	1	1	12	8	79	82	1	2	6	9	123	122	2	-2	1	10	80	79	1
0	4	8	60	66	2	2	12	8	24	24	7	3	6	9	35	36	5	-1	1	10	272	278	2
1	4	8	75	64	3	3	12	8	82	84	5	4	6	9	105	107	1	0	1	10	327	322	4
3	4	8	131	122	1	4	12	8	33	22	14	5	6	9	61	68	7	1	1	10	275	277	4
4	4	8	92	99	4	-3	13	8	39	30	5	6	6	9	19	49	6	2	1	10	78	79	1
5	4	8	77	68	3	-2	13	8	96	96	3	-4	7	9	86	87	3	3	1	10	149	164	4
6	4	8	73	71	4	-1	13	8	50	47	4	-3	7	9	94	94	1	4	1	10	173	183	3
4	5	88	42	38	6	0	13	8	57	45	3	-2	7	9	148	151	1	-4	2	10	167	187	3
3	5	8	25	21	2	1	13	8	46	47	2	-1	7	9	52	55	1	-2	2	10	85	87	1
2	5	8	161	158	1	2	13	8	99	96	2	0	7	9	90	87	2	-3	2	10	179	185	3
1	5	8	136	135	2	3	13	8	33	30	7	1	7	9	52	55	1	-1	2	10	142	143	2
0	5	8	71	70	4	-3	14	8	27	24	5	2	7	9	155	151	2	0	2	10	203	201	2
1	5	10	129	134	2	-2	14	8	24	25	8	3	7	9	95	35	2	1	2	10	138	143	3
2	11	10	68	81	3	5	6	11	60	74	4	2	2	12	114	111	1	2	11	12	51	68	16
3	11	10	115	120	3	-4	7	11	33	24	5	3	2	12	62	62	1	3	11	12	63	17	6
4	11	10	59	58	5	-3	7	11	130	131	2	4	2	12	57	60	3	4	11	12	27	44	10
4	12	10	28	42	11	-2	7	11	27	22	2	5	2	12	44	50	6	-3	12	12	48	44	9
3	12	10	31	37	5	-1	7	11	84	82	1	-4	3	12	93	92	2	-2	12	12	81	75	3
2	12	10	76	78	2	0	7	11	34	23	2	-3	3	12	33	27	2	-1	12	12	49	47	4
1	12	10	100	97	2	1	7	11	85	82	2	-2	3	12	67	65	2	0	12	12	36	32	5
0	12	10	59	57	3	2	7	11	23	23	4	-1	3	12	100	101	1	1	12	12	53	47	4
1	12	10	97	98	1	3	7	11	136	131	2	0	3	12	227	236	2	2	12	12	69	75	3
2	12	10	72	78	4	4	7	11	34	24	8	1	3	12	103	103	3	3	12	12	49	44	5
3	12	10	31	37	7	5	7	11	40	54	6	2	3	12	63	65	2	-2	13	12	74	72	5
4	13	10	56	42	5	-4	8	11	39	37	17	3	3	12	24	27	3	-1	13	12	45	37	9
3	13	10	24	26	6	-3	8	11	34	34	4	4	3	12	88	92	2	0	13	12	16	25	15
2	13	10	34	39	4	-2	8	11	45	41	2	5	3	12	89	104	3	1	13	12	21	37	8
1	13	10	59	51	2	-1	8	11	184	185	4	-4	4	12	59	64	5	2	13	12	62	72	8
0	13	10	43	51	4	0	8	11	34	31	2	-3	4	12	81	82	1	-1	14	12	24	24	9
1	13	10	50	51	2	1	8	11	186	185	1	-1	4	12	66	63	1	0	14	12	48	54	3
2	13	10	28	40	8	2	8	11	40	41	2	-1	4	12	66	67	1	1	14	12	19	24	18
3	13	10	24	26	23	3	8	11	17	34	16	0	4	12	59	63	1	1	0	13	92	83	2

TABLE 7-continued

Observed and calculated structure factors for 2-Mbis-P.

h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s
2	14	10	36	30	4	4	5	4	45	37	6	1	4	12	70	67	1	2	0	13	37	37	2
1	14	10	48	52	3	5	4	12	42	27	11	3	4	12	64	63	3	3	0	13	205	205	2
0	14	10	56	55	3	-4	9	11	40	25	11	3	4	12	79	82	2	4	0	13	48	49	3
1	14	10	39	52	7	-3	9	11	122	123	1	4	4	12	51	64	4	-5	1	13	35	9	10
2	14	10	39	30	5	-2	9	11	88	86	1	5	4	12	28	41	10	-4	1	13	183	182	3
1	15	10	27	18	6	-1	9	11	145	144	1	4	5	12	78	90	8	-3	1	13	88	88	2
0	15	10	29	39	13	0	9	11	41	42	2	-3	5	12	30	33	3	-2	1	13	96	97	3
1	0	11	76	85	2	1	9	11	141	144	4	-2	5	12	115	116	2	0	1	13	164	167	2
2	0	11	48	59	2	2	9	11	88	86	4	-1	5	12	32	29	2	0	1	13	109	106	1
3	0	11	189	196	2	3	9	11	129	123	3	0	5	12	135	139	1	1	1	13	164	167	3
4	0	11	89	91	2	4	9	11	43	45	6	1	5	12	38	29	2	2	1	13	98	97	2
4	1	11	95	94	2	-4	10	11	31	34	13	2	5	12	112	116	2	3	1	13	87	88	1
3	1	11	158	159	4	-3	10	11	85	81	1	3	5	12	30	33	5	4	1	13	169	182	2
2	1	11	64	61	2	-2	10	11	116	109	2	4	5	12	81	90	3	-5	2	13	74	52	5
1	1	11	199	192	4	-1	10	11	32	33	4	5	5	12	16	35	16	-4	2	13	31	32	10
0	1	11	50	42	5	0	10	11	303	302	2	-4	6	12	92	91	4	-3	2	13	44	41	2
1	1	11	194	192	2	1	10	11	36	33	2	-3	6	12	42	45	4	-2	2	13	95	96	2
2	1	11	68	60	1	2	10	11	112	110	5	-2	6	12	38	33	2	-1	2	13	141	145	1
3	1	11	155	158	4	3	10	11	77	80	3	-1	6	12	85	86	1	0	2	13	68	61	1
4	1	11	92	94	2	4	10	11	35	34	7	0	6	12	116	120	1	1	2	13	140	145	1
4	2	11	100	104	3	-4	11	11	43	48	6	1	6	12	86	85	1	2	2	13	101	96	1
3	2	11	23	14	10	-3	11	11	40	42	3	2	6	12	33	34	2	3	2	13	43	41	2
2	2	11	98	95	2	-2	11	11	70	67	2	3	6	12	50	45	4	4	2	13	38	32	4
1	2	11	279	296	4	-1	11	11	107	103	1	4	6	12	87	91	3	5	2	13	29	31	13
0	2	11	183	186	8	0	11	11	24	4	11	5	6	12	13	30	12	-4	3	13	61	61	4
2	2	11	97	95	1	2	11	11	66	67	3	-3	7	12	54	48	10	-3	3	13	35	22	11
3	2	11	17	15	5	3	11	11	33	42	21	-2	7	12	162	158	1	-1	3	13	88	89	1
4	2	11	102	104	2	4	11	11	39	47	6	-1	7	12	110	108	1	0	3	13	163	158	1
3	3	11	117	117	2	-2	12	11	76	77	5	1	7	12	111	108	1	2	3	13	96	96	2
2	3	11	46	48	3	-1	12	11	66	64	3	2	7	12	154	158	4	3	3	13	24	21	3
1	3	11	23	29	2	0	12	11	25	18	10	3	7	12	78	78	3	4	3	13	51	62	4
0	3	11	128	145	3	1	12	11	62	64	3	4	7	12	44	48	5	5	3	13	54	62	5
1	3	11	25	20	2	2	12	11	77	77	10	5	7	12	25	32	10	-4	4	13	76	80	6
2	3	11	47	47	1	3	12	11	22	18	18	-4	8	12	68	72	4	-3	4	13	56	49	2
3	3	11	112	117	3	-3	13	11	49	56	6	-3	8	12	71	67	2	-2	4	13	55	57	1
4	3	11	73	74	3	-2	13	11	53	55	4	-2	8	12	87	84	1	-1	4	13	75	71	3
5	3	11	45	52	5	-1	13	11	27	20	6	-1	8	12	90	89	1	0	4	13	18	19	3
4	4	11	65	66	2	0	13	11	50	40	6	0	8	12	88	60	2	1	4	13	77	72	2
3	4	11	189	186	3	1	13	11	34	20	8	1	8	12	90	89	1	2	4	13	59	58	2
2	4	11	26	24	3	2	13	11	40	55	12	2	8	12	85	84	2	3	4	13	54	49	2
1	4	11	101	100	1	3	13	11	55	56	7	3	8	12	72	67	4	4	4	13	72	80	3
0	4	11	79	83	1	-2	14	11	25	13	10	4	8	12	66	72	5	5	4	13	24	32	11
1	4	11	101	100	1	-1	14	11	26	28	10	5	8	12	13	36	12	-4	5	13	37	38	4
2	4	11	26	24	5	0	14	11	36	41	4	-4	9	12	84	92	2	-3	5	13	80	79	2

TABLE 7-continued

Observed and calculated structure factors for 2-Mbis-P

h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	h	k	l	10Fo	10Fc	10s	
3	4	11	185	286	3	2	14	11	18	114	2	-2	5	13	81	79	1	5	1	14	4	28	4	
4	4	11	63	66	3	1	14	11	27	14	4	-1	5	13	26	26	2	-5	2	14	4	76	84	
5	4	11	47	52	5	0	0	12	640	627	11	0	5	13	266	274	2	-4	2	14	75	78	2	
4	5	11	17	12	8	2	0	12	27	27	2	0	5	13	32	26	4	-3	2	14	215	215	7	
3	5	11	106	101	4	2	0	12	153	135	1	1	5	13	84	79	2	-1	2	14	79	79	2	
2	5	11	98	100	1	3	0	12	93	99	3	2	5	13	78	79	3	-2	2	14	152	146	1	
1	5	11	136	133	2	4	0	12	110	113	2	3	4	5	39	38	6	0	2	14	19	1	4	
0	5	11	78	75	1	-4	1	12	191	207	3	4	5	13	53	60	5	1	2	14	151	146	3	
1	5	11	134	133	2	-3	1	12	150	158	3	-4	6	13	81	88	8	2	2	14	84	79	2	
2	5	11	94	101	2	-2	1	12	42	37	2	-3	6	13	44	36	5	3	2	14	222	215	1	
3	5	11	105	101	2	-1	1	12	132	137	2	-2	6	13	111	110	1	4	2	14	73	78	2	
4	5	11	30	12	6	0	1	12	130	121	1	-1	6	13	143	140	1	5	2	14	54	84	5	
5	5	11	46	42	6	1	1	12	131	137	1	0	6	13	23	11	3	-5	3	14	52	44	6	
4	6	11	74	73	2	2	1	12	38	37	3	1	6	13	142	140	2	-4	3	14	33	39	5	
3	6	11	63	60	3	3	1	12	145	157	4	2	6	13	112	110	1	-3	3	14	49	48	2	
2	6	11	34	33	3	4	1	12	197	207	3	3	6	13	47	36	5	-2	3	14	57	50	3	
1	6	11	141	145	1	-4	2	12	60	60	2	4	6	13	76	88	3	-1	3	14	83	81	1	
0	6	11	47	30	3	-3	2	12	60	61	2	-3	6	13	24	44	11	0	3	14	63	65	1	
1	6	11	142	145	2	-2	2	12	115	112	1	-2	7	13	55	64	3	1	3	14	84	82	3	
2	6	11	32	33	2	-1	2	12	192	198	2	-1	7	13	100	95	2	2	3	14	54	50	1	
3	6	11	60	60	3	0	2	12	104	106	1	0	7	13	110	107	2	3	3	14	53	48	3	
4	6	11	66	72	4	1	2	12	195	198	2	1	7	13	80	80	2	4	3	14	31	39	8	
5	3	14	26	44	11	2	1	15	160	151	2	1	10	16	73	78	4	1	0	18	129	120	2	
4	4	14	46	22	6	3	1	15	136	131	1	2	10	16	25	23	6	2	0	18	197	195	2	
4	4	14	23	12	8	4	1	15	49	57	5	-1	10	16	25	13	7	3	0	18	23	20	11	
2	4	14	34	19	3	-4	2	15	35	56	11	0	10	16	21	23	9	-4	1	18	48	50	4	
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2	6	14	104	102	1	3	3	15	33	36	3	-1	2	17	35	33	4	-3	3	18	25	5	6	
1	6	14	42	41	2	4	3	15	40	46	6	0	2	17	42	22	4	-2	3	18	63	57	2	

TABLE 7-continued

Observed and calculated structure factors for 2-Mbis-P.

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4	7	14	73	82	9	-1	4	15	24	16	3	-4	3	16	64	59	2	3	2	17	23	22	7
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TABLE 7-continued

Observed and calculated structure factors for 2-Mbis-P

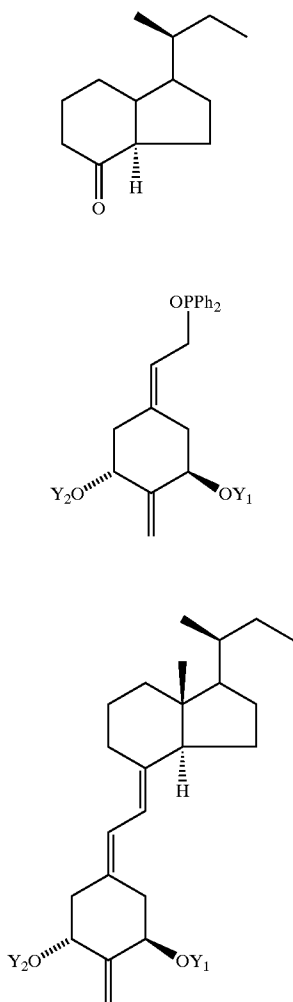
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0	7	19	27	20	15	-2	2	20	72	71	2	-1	5	20	27	20	6	0	2	21	28	24	5	
1	7	19	32	23	5	-1	2	20	40	41	3	0	5	20	18	18	17	1	1	21	45	47	5	

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Synthesis of 2MBP

EXAMPLE 3

The preparation of 2MBP having the basic structure I can be accomplished by a common general method, i.e. the condensation of a bicyclic Windaus-Grundmann type ketone II with the allylic phosphine oxide III to the corresponding 2-methylene-19-nor-vitamin D analog IV followed by deprotection at C-1 and C-3 in the latter compound IV to obtain compound I, i.e. 2MBP.



In the structures II, III, and IV groups Y₁ and Y₂ are hydroxy-protecting groups, it being also understood that any functionalities that might be sensitive, or that interfere with the condensation reaction, be suitably protected as is well-known in the art. The process shown above represents an application of the convergent synthesis concept, which has been applied effectively for the preparation of vitamin D compounds [e.g. Lythgoe et al., *J. Chem. Soc. Perkin Trans. I*, 590 (1978); Lythgoe, *Chem. Soc. Rev.* 9, 449 (1983); Toh et al., *J. Org. Chem.* 48, 1414 (1983); Baggiolini et al., *J. Org. Chem.* 51, 3098 (1986); Sardina et al., *J. Org. Chem.* 51, 1264 (1986); *J. Org. Chem.* 51, 1269 (1986); DeLuca et al., U.S. Pat. No. 5,086,191; DeLuca et al., U.S. Pat. No. 5,536,713].

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Hydrindanones of the general structure II are known, or can be prepared by known methods.

For the preparation of the required phosphine oxides of general structure III, a synthetic route has been developed starting from a methyl quinate derivative which is easily obtained from commercial (1R,3R,4S,5R)-(-)-quinic acid as described by Perlman et al., *Tetrahedron Lett.* 32, 7663 (1991) and DeLuca et al., U.S. Pat. No. 5,086,191.

The overall process of the synthesis of compound I is illustrated and described more completely in U.S. Pat. No. 5,843,928 entitled "2-Alkylidene-19-Nor-Vitamin D Compounds" and in application Ser. No. 09/878,438 filed Jun. 11, 2001 entitled "1 α -Hydroxy-2-Methylene-19-Nor-Homopregnacalciferol and its Uses" the specifications of which are specifically incorporated herein by reference.

As used in the description and in the claims, the term "hydroxy-protecting group" signifies any group commonly used for the temporary protection of hydroxy functions, such as for example, alkoxy carbonyl, acyl, alkylsilyl or alkylarylsilyl groups (hereinafter referred to simply as "silyl" groups), and alkoxyalkyl groups. Alkoxy carbonyl protecting groups are alkyl-O-CO- groupings such as methoxycarbonyl, ethoxycarbonyl, propoxycarbonyl, isopropoxycarbonyl, butoxycarbonyl, isobutoxycarbonyl, tert-butoxycarbonyl, benzyloxycarbonyl or allyloxycarbonyl. The term "acyl" signifies an alkanoyl group of 1 to 6 carbons, in all of its isomeric forms, or a carboxyalkanoyl group of 1 to 6 carbons, such as an oxalyl, malonyl, succinyl, glutaryl group, or an aromatic acyl group such as benzoyl, or a halo, nitro or alkyl substituted benzoyl group. The word "alkyl" as used in the description of the claims, denotes a straight-chain or branched alkyl radical of 1 to 10 carbons, in all its isomeric forms. Alkoxyalkyl protecting groups are groupings such as methoxymethyl, ethoxymethyl, methoxyethoxymethyl, or tetrahydrofuranyl and tetrahydropyranyl. Preferred silyl-protecting groups are trimethylsilyl, triethylsilyl, t-butyl dimethylsilyl, dibutylmethylsilyl, diphenylmethylsilyl, phenyldimethylsilyl, diphenyl-t-butylsilyl and analogous alkylated silyl radicals. The term "aryl" specifies a phenyl-, or an alkyl-, nitro or halo-substituted phenyl group.

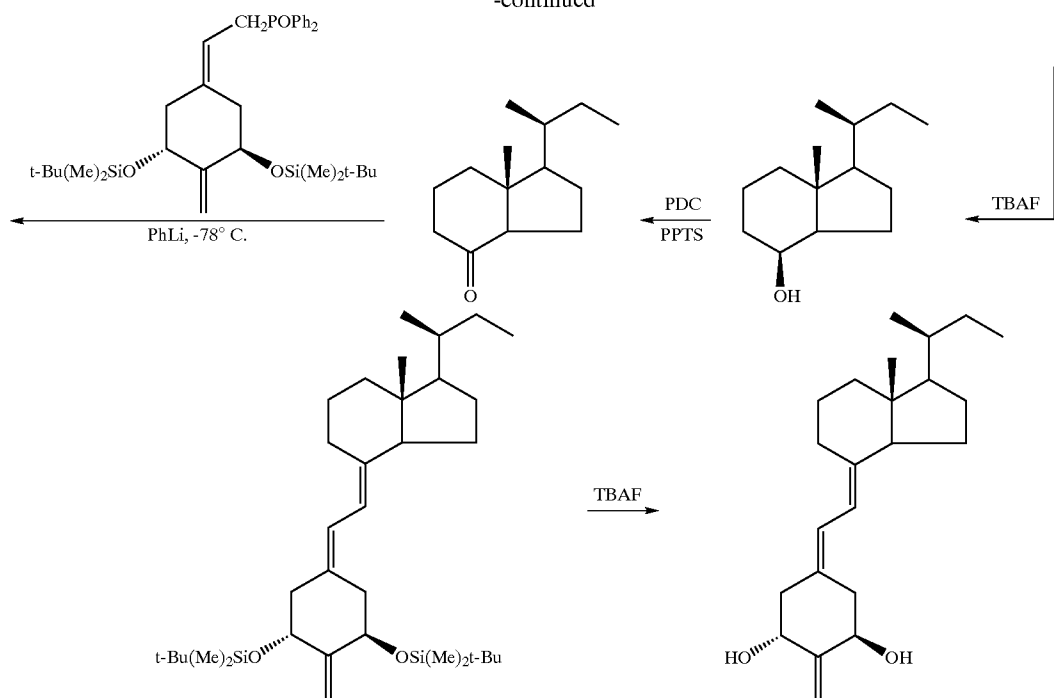
A "protected hydroxy" group is a hydroxy group derivatized or protected by any of the above groups commonly used for the temporary or permanent protection of hydroxy functions, e.g. the silyl, alkoxyalkyl, acyl or alkoxy carbonyl groups, as previously defined. The terms "hydroxyalkyl", "deuteroalkyl", "aminoalkyl", "halogenalkyl", "alkoxyalkyl", "aryloxyalkyl", and "fluoroalkyl" refer to an alkyl radical substituted by one or more hydroxy, deuterium, amino, halogen, alkoxy, aryloxy, or fluoro group respectively. A "halogen" group includes any of the five elements fluorine, chlorine, bromine, iodine and astatine that form a part of group VIIA of the periodic table.

Referring now to Schemes I-III, there are illustrated three different methods of synthesizing 2MBP starting with vitamin D₂. The first five steps are identical (shown in Scheme I) for each synthesis to obtain the protected 20(S)-aldehyde. Thereafter, in Schemes I and II the aldehyde is converted to a protected 20(S)-alcohol which in turn has its side chain converted to bis-homo in two different ways (Scheme I versus Scheme II). In contrast, Scheme III illustrates a direct conversion of the protected 20(S)-aldehyde to the bis-homo side chain.

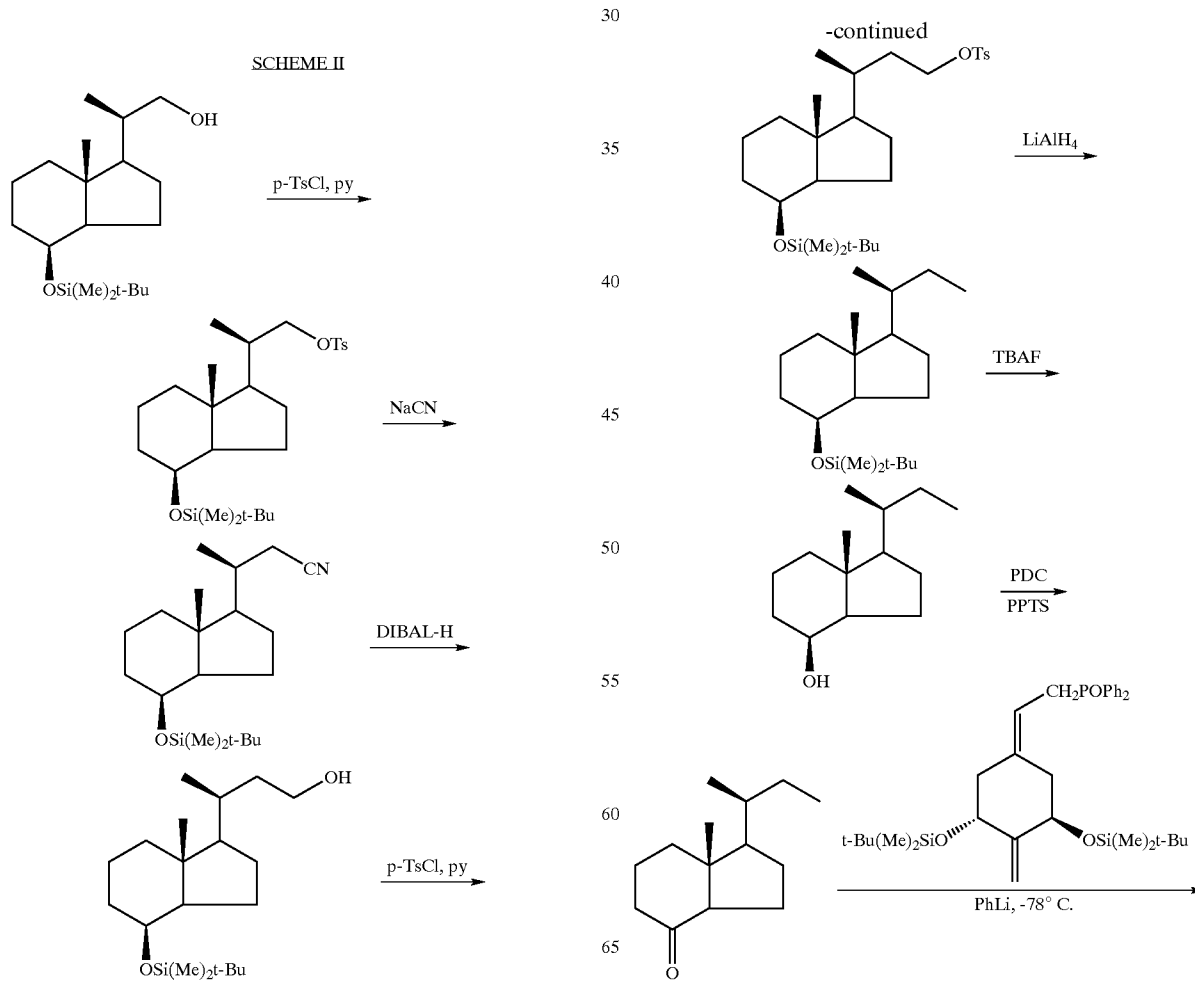
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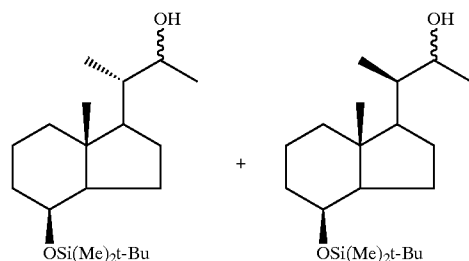
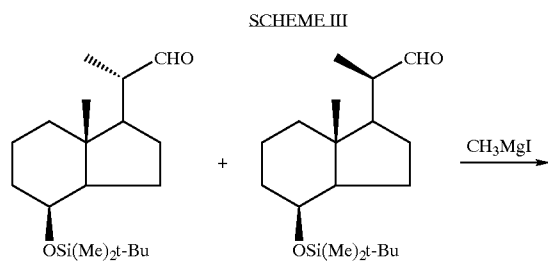
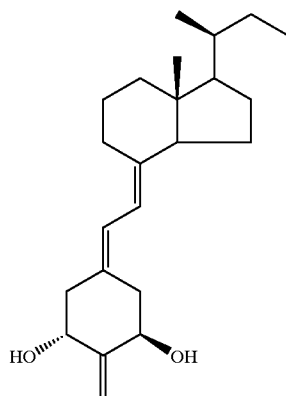
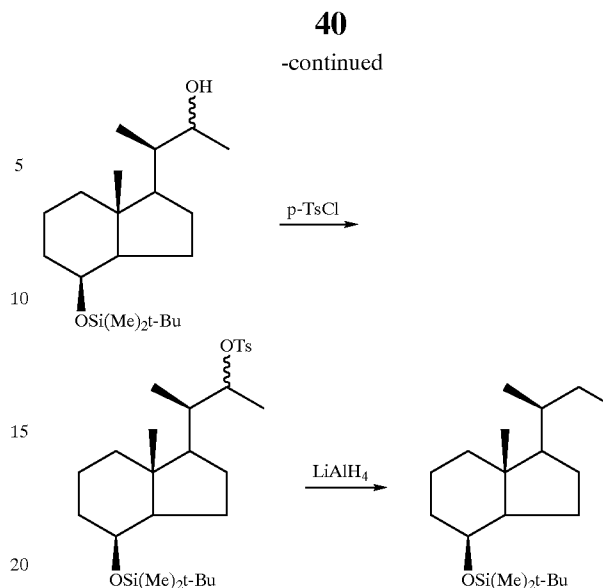
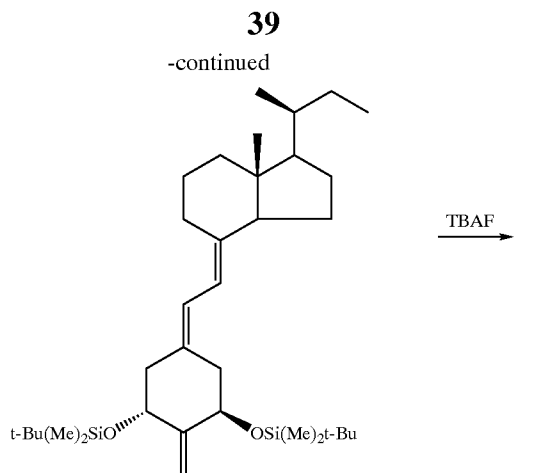
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SCHEME II





We claim:

1. (20S)-1 α -hydroxy-2-methylene-19-nor-bishomopregnacalciferol in crystalline form.
2. The crystalline form of (20S)-1 α -hydroxy-2-methylene-19-nor-bishomopregnacalciferol having molecular packing arrangement defined by orthorhombic space group P2(1)2(1)2(1) and unit cell dimensions a=6.6 \AA ⁰, b=15.5 \AA ⁰, c=20.1 \AA ⁰, $\alpha=90^\circ$, $\beta=90^\circ$ and $\gamma=90^\circ$.
3. A three dimensional structure for (20S)-1 α -hydroxy-2-methylene-19-nor-bishomopregnacalciferol as defined by the molecular packing arrangement set forth in claim 2.
4. A method of purifying (20S)-1 α -hydroxy-2-methylene-19-nor-bishomopregnacalciferol, comprising the steps of:
 - (a) boiling a solvent selected from the group consisting of acetone, ethyl acetate, isopropanol, chloroform, dichloromethane and diethyl ether under inert atmosphere;
 - (b) dissolving a product containing (20S)-1 α -hydroxy-2-methylene-19-nor-bishomopregnacalciferol to be purified in said solvent;
 - (c) cooling said solvent and dissolved product below ambient temperature for a sufficient amount of time to form a precipitate of (20S)-1 α -hydroxy-2-methylene-19-nor-bishomopregnacalciferol crystals, and
 - (d) separating the (20S)-1 α -hydroxy-2-methylene-19-nor-bishomopregnacalciferol crystals from the solvent.
5. The method of claim 4 including the further step of allowing said solvent and dissolved product to cool to ambient temperature prior to cooling below ambient temperature.
6. The method of claim 4 wherein said inert atmosphere is an argon atmosphere.
7. The method of claim 4 wherein said solvent and dissolved product is cooled to between about 35° F. to about 45° F.
8. The method of claim 4 wherein the step of separating comprises filtering the solvent and precipitate to obtain the crystals.
9. The method of claim 4 including a further step (e) comprising repeating steps (a) through (d) using the recovered crystals from step (d) as the product of step (b).

* * * * *